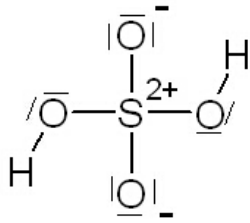
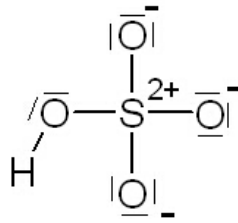


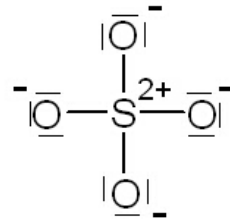
# Komplexanionen aus Säuren



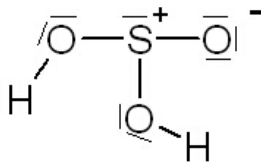
Schwefelsäure  $\text{H}_2\text{SO}_4$



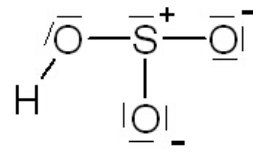
Hydrogensulfat  $\text{HSO}_4^-$



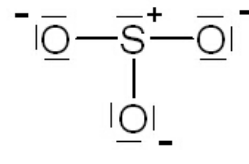
Sulfat  $\text{SO}_4^{2-}$



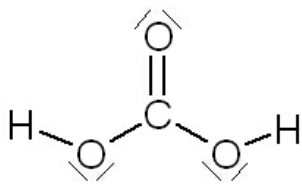
Schweflige Säure  $\text{H}_2\text{SO}_3$



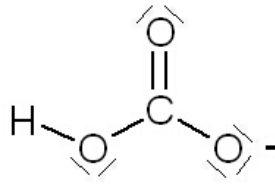
Hydrogensulfit  $\text{HSO}_3^-$



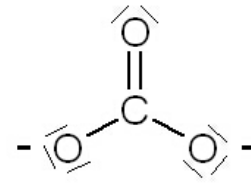
Sulfit  $\text{SO}_3^{2-}$



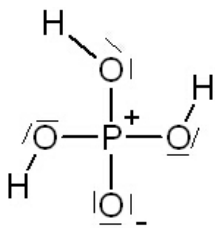
Kohlensäure  $\text{H}_2\text{CO}_3$



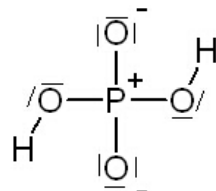
Hydrogencarbonat  $\text{HCO}_3^-$



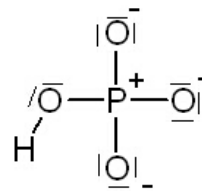
Carbonat  $\text{CO}_3^{2-}$   
(mesomer)



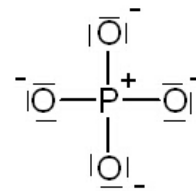
Phosphorsäure  $\text{H}_3\text{PO}_4$



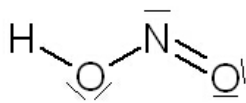
Dihydrogenphosphat  $\text{H}_2\text{PO}_4^-$



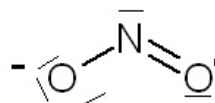
Hydrogenphosphat  $\text{HPO}_4^{2-}$



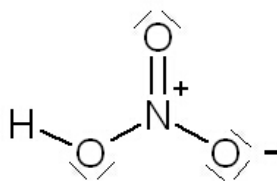
Phosphat  $\text{PO}_4^{3-}$



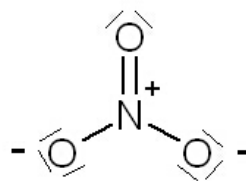
Salpetrige Säure  $\text{HNO}_2$



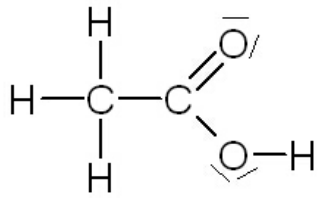
Nitrit  $\text{NO}_2^-$   
(mesomer)



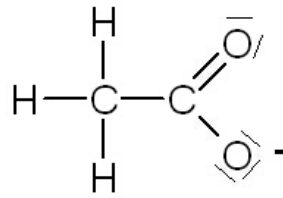
Salpetersäure  $\text{HNO}_3$



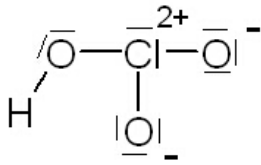
Nitrat  $\text{NO}_3^-$   
(mesomer)



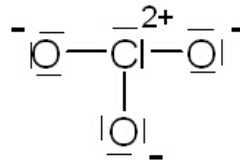
Essigsäure  $\text{CH}_3\text{COOH}$



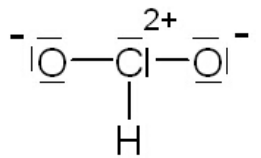
Acetat  $\text{H}_3\text{COO}^-$   
(mesomer O-C-O)



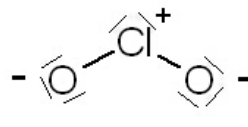
Chlorsäure  $\text{HClO}_3$



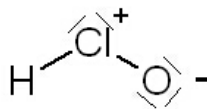
Chlorat  $\text{ClO}_3^-$



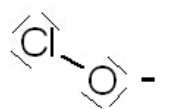
Chlorige Säure  $\text{HClO}_2$



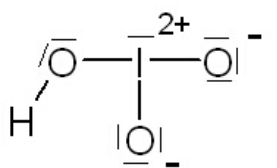
Chlorit  $\text{ClO}_2^-$



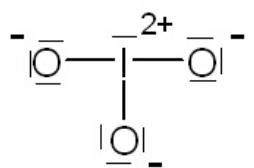
Hypochlorige Säure  $\text{HClO}$   
Javellewasser



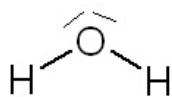
Hypochlorit  $\text{ClO}^-$



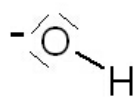
Iodsäure  $\text{HIO}_3$



Iodat  $\text{IO}_3^-$

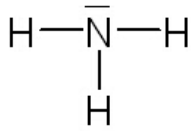


Wasser  $\text{H}_2\text{O}$

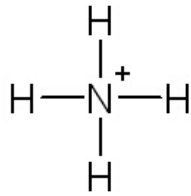


Hydroxid  $\text{OH}^-$

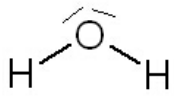
# Komplekationen aus Basen



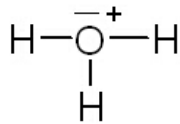
Ammoniak  $\text{NH}_3$



Ammonium  $\text{NH}_4^+$



Wasser  $\text{H}_2\text{O}$



Hydronium  $\text{H}_3\text{O}^+$